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Acids Bases  
Arrhenius  
Bronsted acid. A

substance capable of donating a proton (proton donor)

Bronsted base. A substance that can accept a proton (proton acceptor) Conjugate acid-base pair.

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Aqueous solutions

acids have a (blank)

taste. Sour Taste. What

color does litmus paper

turn when it's dipped in

an acid. Red. What

color does litmus paper

turn when it's dipped in

a base.

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**Chemistry Chapter  
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React Flashcards ...**

Chapter 15 Acids and Bases. strong acid. strong base. weak acid. weak base. an acid that ionizes completely in solvent. a base that ionizes completely in a solvent. an acid that releases few hydrogen ions in aqueous solution. a base that releases few hydroxide ions in aqueous solution.



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Chapter 15: Acids and  
Bases Acids and Bases

Arrhenius Definitions:

acids - compounds that  
produce an increase in  
[H<sup>+</sup>] when dissolved in

water bases -

compounds that  
produce an increase in  
[OH<sup>-</sup>] when dissolved

in water Lewis

Definitions: acids -

electron pair acceptors

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Acids Bases  
Answers  
bases - electron pair  
donors Brønsted-Lowry  
Definitions:

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Define acid and base

accord-ing to

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### Acids Bases

Arrhenius's theory of ionization. Explain the differences between strong and weak acids and bases. FIGURE 15-1 (a) Fruits and fruit juices contain acids such as citric acid and ascorbic acid. Carbonated beverages contain benzoic acid, phosphoric acid, and carbonic acid. (b) Many household cleaners contain ...

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Chapter 15 Aqueous Equilibria: Acids and Bases. Instructor's Resource Materials (Download only) for Chemistry, ...

Conjugate Acid-Base Pairs: Chemical species whose formulas differ only by one hydrogen ion,  $H^+$  Brønsted-Lowry Acid: A substance that can transfer hydrogen ions,  $H^+$ . In other words, a proton donor

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### Acids Bases

#### **Chapter 15**

#### **Equilibria: Acids and Bases**

Sec 15.1: Heartburn:

Sec 15.2: The Nature of Acids and Bases:

Properties of Acids-sour taste, ability to

dissolve many metals, ability to turn blue

litmus paper red, and ability to neutralize

bases. Properties of

Bases-bitter taste,

slippery feel, ability to

turn red litmus paper

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blue, and ability to neutralize acids.

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Period Honors

Chemistry — Chapter  
15 Acid & Base Test.

Matching: L1. strong acids A. bitter L2. pH B. ionize only slightly in solution £3. titration C. salt + H<sub>2</sub> (g) F 4. neutralization D. salt + ...

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Chapter 15 Acids and  
Bases. Practice  
Exercise 15.11;

Comments on my  
teaching methods.

What works for you?  
Please be honest!

Chapter 16 Acid-Base  
Equilibria & Solubility  
Equilibria. CHAPTER 16  
TEST

ANNOUNCEMENT!!!!!!

Quick Chapter 16  
notes; Chapter 16  
Homework 11/30;



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Homework due 12/6;  
Homework due 12/07;  
Homework due 12/10;  
In Class problem 12 ...

## **Baylor, Scott / Chapter 15 Acids and Bases**

Chapter 15 Test Acids  
And Bases Answers  
when acids or bases  
act as acids or bases,  
they form their  
conjugate and the  
strength is the inverse  
of their parent. and  
there are certain

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strong acids and strong bases. LiOH, NaOH, KOH - when act as bases (get rid of H<sup>+</sup> ion) then left with Li  
Page 13/24

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and Solubility Equilibria

Chapter 17. Entropy,

Free Energy and

Equilibrium Chapter 18

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Chapter Test Name \_\_\_\_\_  
Class Date \_\_\_\_\_ Chapter  
Test A, continued

\_\_\_\_ 14. A strong base in an aqueous solution  
a. is a weak electrolyte. b. produces many H ions. c. will not dissolve. d. completely dissociates into ions.

\_\_\_\_ 15. In a Brønsted-Lowry acid-base reaction, what are transferred from one reactant to another? a. electrons ...

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